



CET25C10 BIOMOLECULES

Class 12 - Chemistry

Time Allowed: 1 hour and 30 minutes

Maximum Marks: 75

1. The carrier of hereditary character is. [1]
 - a) Lipids
 - b) Cytochromes
 - c) Nucleotides
 - d) Nucleosides
2. In the ring structure of glucose, the anomeric carbon is: [1]
 - a) C-4
 - b) C-3
 - c) C-1
 - d) C-2
3. DNA fingerprinting is used [1]
 - a) to determine paternity of an individual
 - b) in forensic laboratories
 - c) all of these
 - d) to identify racial groups
4. The vitamins are generally stored in the body in the [1]
 - a) abdomen
 - b) muscles
 - c) liver and adipose tissue
 - d) pancreas
5. An example of globular protein is [1]
 - a) Collagen
 - b) Primary protease
 - c) Histones
 - d) Albumin
6. Peptide linkage is present in: [1]
 - a) Carbohydrates
 - b) Proteins
 - c) Vitamins
 - d) Rubber
7. A vitamin which plays a vital role in the clotting of blood is: [1]
 - a) Vitamin K
 - b) Vitamin D
 - c) Vitamin B
 - d) Vitamin A
8. The helix structure of proteins is stabilized by: [1]
 - a) disulphide bond
 - b) peptide bond
 - c) van der Waals forces
 - d) hydrogen bond
9. Which of the following vitamins is water soluble? [1]
 - a) Vitamin D
 - b) Vitamin C
 - c) Vitamin A
 - d) Vitamin E

- a) tertiary structure
b) quaternary structure
c) secondary structure
d) primary structure
20. The action of nitrous acid on ethylamine gives mainly: [1]
a) ethyl alcohol
b) ethyl nitrite
c) nitroethane
d) ethane
21. The sugar constituent of DNA is [1]
a) D – ribose
b) D – 2 – deoxy ribose
c) D – glucose
d) D – lactose
22. Which of the following is a disaccharide? [1]
a) Glucose
b) Cellulose
c) Lactose
d) Starch
23. β -pleated sheet structure in proteins refers to [1]
a) tertiary structure
b) primary structure
c) quaternary structure
d) secondary structure
24. Which parts of amino acids molecules are linked through hydrogen bonds in the secondary structure of proteins? [1]
a) $\begin{array}{c} \text{—C—} \\ || \\ \text{O} \end{array}$ and -NH- groups
b) COOH group
c) NH_2 group
d) None of these
25. One or more of the following vitamin is insoluble in water [1]
a) all of these
b) vitamin D
c) vitamin K
d) vitamin E
26. Glucose and fructose are: [1]
a) enantiomers of each other.
b) anomers of each other.
c) isomers of each other.
d) Homologous of each other.
27. The deficiency of which of the following vitamins causes Scurvy? [1]
a) Vitamin A
b) Vitamin B₆
c) Vitamin B₁₂
d) Vitamin C
28. Nucleic acids are the polymers of [1]
a) sugars
b) bases
c) nucleosides
d) nucleotides
29. Which of the following bases is not present in DNA? [1]
a) Adenine
b) Cytosine
c) Uracil
d) Thymine
30. Which of the following B group vitamins can be stored in our body? [1]
a)
b)

- Vitamin B₂ Vitamin B₁
- c) Vitamin B₆ d) Vitamin B₁₂
31. Amino acids are: [1]
- a) Neutral b) Acidic
- c) Basic d) Amphoteric
32. Curdling of milk which is caused due to [1]
- a) formation of lactic acid by the bacteria present in milk and resulting denaturation b) fall in temperature
- c) increase in pH d) disturbance in primary structure of milk proteins
33. Deficiency of which of the following vitamins causes Pernicious anaemia? [1]
- a) Vitamin B₂ b) Vitamin B₁
- c) Vitamin B₁₂ d) Vitamin B₆
34. Nucleotides are joined together by [1]
- a) peptide linkage b) disulphide linkage
- c) glycosidic linkage d) phosphodiester linkage
35. An α -helix is a structural feature of: [1]
- a) Starch b) Polypeptides
- c) Nucleotides d) Sucrose
36. Cheilosis is caused by deficiency of [1]
- a) Vitamin B₆ b) Vitamin B₂
- c) Vitamin C d) Vitamin B₁₂
37. ____ are joined together by phosphodiester linkage between 5' and 3' carbon atoms of the pentose sugar. [1]
- a) Nucleosides b) Nucleic acids
- c) Proteins d) Nucleotides
38. Which one is the complementary base of cytosine in one strand to that in other strand of DNA? [1]
- a) Uracil b) Thymine
- c) Guanine d) Adenine
39. The deficiency of which of the following vitamins causes **Rickets**? [1]
- a) Vitamin B b) Vitamin A
- c) Vitamin D d) Vitamin C
40. RNA guides the biosynthesis of [1]
- a) Cellulose b) Fats
- c) Starch d) Proteins

41. Water soluble vitamins must be supplied regularly in diet because [1]
a) they cannot be provided by synthetic means. b) they are not widely available.
c) they are readily excreted in urine and cannot be stored (except vitamin B₁₂) in our body. d) they get used up very fast in body.
42. Adenosine is [1]
a) Nucleic acid b) Nucleoside
c) Base d) Nucleotide
43. Proteins are polymers of [1]
a) Monosaccharides b) Amino acids
c) Nucleic acids d) Amines
44. Hydrolysis of sucrose is called [1]
a) esterification b) saponification
c) inversion d) hydration
45. On hydrolysis, which of the following carbohydrates gives only glucose ? [1]
a) Galactose b) Maltose
c) Lactose d) Sucrose
46. Protein is a [1]
a) Addition polymer b) Copolymer
c) Homopolymer d) Condensation polymer
47. Commercially glucose is obtained by [1]
a) hydrolysis of sucrose b) boiling sucrose with dilute HCl or H₂SO₄ in alcoholic solution
c) hydrolysis of starch by boiling it with dilute H₂SO₄ at 393 K under pressure. d) crushing ripe grapes
48. The vitamin that cannot be stored in our body is [1]
a) vitamin E b) vitamin C
c) vitamin K d) vitamin D
49. शॉक थेरेपी का मॉडल किस पर लागू किया गया? [1]
a) पूँजीवादी देशों पर b) साम्यवादी देशों पर
c) भिन्न-राष्ट्रों पर d) धुरी-राष्ट्रों पर
50. When a nucleoside is linked to phosphoric acid which of the following may be obtained? [1]
a) Ribonucleic acid b) Nucleotide
c) An amino acid d) Deoxyribonucleic acid
51. Which of the following reactions of glucose can be explained only by its cyclic structure? [1]
a) Glucose is oxidised by nitric acid to b) Pentaacetate of glucose does not react with

- gluconic acid. hydroxylamine.
- c) Glucose reacts with hydroxylamine to form an oxime. d) Glucose forms pentaacetate.
52. On hydrolysis, which of the following carbohydrates gives glucose and galactose? [1]
- a) Maltose b) Lactose
c) Sucrose d) Cellulose
53. On hydrolysis, which of the following carbohydrates gives glucose and fructose? [1]
- a) Starch b) Maltose
c) Lactose d) Sucrose
54. The major component of starch is: [1]
- a) amylopectin b) water
c) amylose d) glucose
55. Which of the following statements is not true about glucose? [1]
- a) On heating with HI it forms n-hexane. b) It gives 2, 4 DNP test.
c) It is an aldohexose. d) It is present in pyranose form.
56. The general formula for carbohydrate is: [1]
- a) $C_x(H_2O)_x$ b) $C_{x-1}(H_2O)_{2y}$
c) $C_{x+1}(H_2O)_y$ d) $C_{2x}(H_2O)_y$
57. Dinucleotide is obtained by joining two nucleotides together by a phosphodiester linkage. Between which carbon atoms of pentose sugars of nucleotides are these linkages present? [1]
- a) 5' and 3' b) 3' and 3'
c) 1' and 5' d) 5' and 5'
58. Amino acid is. [1]
- a) $H_2N.CH_2.COOH$ b) $Cl - CH_2.COOH$
c) $HO.CH_2COOH$ d) CH_3COONH_4
59. The glycosidic linkage involved in linking the glucose units in amylase part of starch is: [1]
- a) $C_1 - C_4 \beta$ linkage b) $C_1 - C_6 \beta$ linkage
c) $C_1 - C_4 \alpha$ linkage d) $C_1 - C_6 \alpha$ linkage
60. Progesterone is responsible for [1]
- a) preparing the uterus for implantation of fertilised egg. b) development of secondary female characteristics.
c) controlling menstrual cycle. d) development of secondary male characteristics.
61. The biological polymer is [1]

- a) Stearic acid
b) Nucleic acid
c) Palmitic acid
d) Formic acid

62. The most appropriate base pair in a double helix of DNA is [1]

- a) A – T
b) A – C
c) C – T
d) T – G

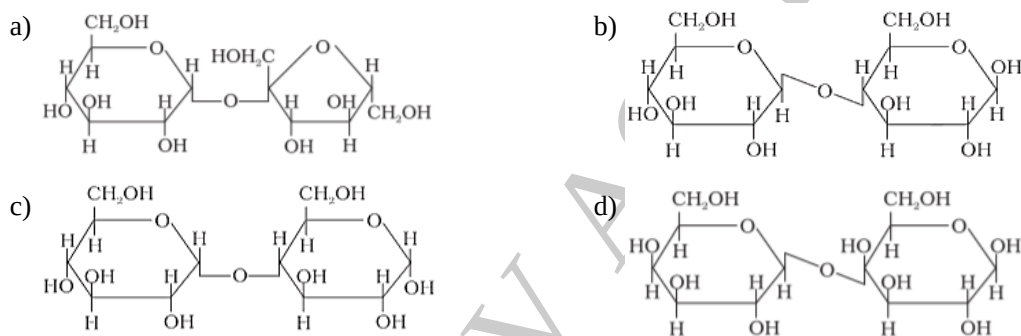
63. Building unit of a protein is [1]

- a) β – Aminoacid
b) λ – Aminoacid
c) γ – Aminoacid
d) α – Aminoacid

64. DNA fingerprinting means the [1]

- a) sequencing the nucleotides in DNA
b) sequencing the bases present in double helix
c) information regarding the unique sequence of bases on DNA for a person
d) information regarding the unique imprints on the fingertip for a person

65. In disaccharides, if the reducing groups of monosaccharides i.e. aldehydic or ketonic groups are bonded, these are non-reducing sugars. Which of the following disaccharide is a non-reducing sugar? [1]



66. α -D(+) glucose and β -D(+) glucose are [1]

- a) Anomers
b) Enantiomers
c) Optical isomers
d) Geometrical isomers

67. Which of the following acids is a vitamin? [1]

- a) Adipic acid
b) Aspartic acid
c) Saccharic acid
d) Ascorbic acid

68. Zwitter ion is represented as: [1]



69. Which of the following is a fibrous protein? [1]

- a) Glycoprotein
b) Keratin
c) Proteases
d) Prolamine

70. Glucose is: [1]
- a) Aldopentose
 - b) Ketopentose
 - c) Aldohexose
 - d) Ketohexose
71. Which one of the following is not a globular protein? [1]
- a) Insulin
 - b) Enzyme
 - c) Haemoglobin
 - d) Myosin
72. Which one is not the essential amino acid in the ones given below? [1]
- a) Valine
 - b) Proline
 - c) Leucine
 - d) Arginine
73. When D-glucose reacts with HI, it forms [1]
- a) Iodohexane
 - b) Saccharic acid
 - c) Gluconic acid
 - d) n-hexane
74. Certain vitamins cannot be stored in the body because [1]
- a) they are readily excreted in urine
 - b) they are used up very fast in the body
 - c) they are insoluble in water
 - d) they are soluble in fat
75. Maltose is made of: [1]
- a) α -D-glucose
 - b) α - D-glucose and β - D -glucose
 - c) Glucose and fructose
 - d) D-fructose